MODERN PETROL AND CLASSIC CARS - THE MANCHESTER XPAG TESTS

By Paul Ireland

Even if your classic car runs well on modern fuel, the chances are you know somebody who has problems. The most common issue people suffer from is called the "Hot Restart Problem". Drive your car any further than 10 miles or so, stop for 10 minutes, for example to fill up with petrol, and when you get back your car it will not start. A related problem occurs if you are driving in slow traffic, especially on a warm day, the engine coughs and splutters to a stop as though it has run out of fuel.

These are the most obvious problems with modern fuel. There are others which people have reported, including burned exhaust valves, cracked cylinder heads, not to mention the worries of ethanol blended fuels.

I have owned my MG TC since 1967 and, for those who can remember back that far, used to run it on two-star leaded fuel. It ran like a dream. My problems started after the demise of leaded petrol and have resulted in a great deal of time spent in the garage trying to sort them out.

Around 15 years ago, I realised my problems were caused by differences in the composition of modern fuel and started testing different concoctions and types of petrol in my TC, writing articles about my findings. These tests culminated in a student engineering project at the School of Mechanical, Aerospace and Civil Engineering (MACE) who mounted an XPAG engine similar to the one fitted in my TC in an

engine test cell aimed at investigating these problems. Unfortunately, the students ran out of time before being able to perform a full investigation. The good news is that with additional support from the MG Car Club and help from MACE, these tests have been completed, providing a fuller understanding of the problems caused by modern fuel, things that can be done to mitigate them and, on the way, debunked a few myths.

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Oust who loaned the engine, David
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Why an XPAG?

Almost the first thing people say is "why test an XPAG, they are an old engine, designed in the late 1930's and only fitted to MG T Types". While it would have been ideal to test a range of engines, the high cost of installing the engine in the test cell prevented this. In practice, the XPAG or 'X' series engines were used in virtually all Morris and Wolseley cars until 1956, including the many thousands of the

Morris 10/4 Utility cars and vans made during WW2.

The XPAG is a good compromise. Its long stroke bottom end shares a great deal with earlier engines, while the cylinder head design is virtually identical to the A and B series engines fitted to later cars. It also demonstrates the problems of running with modern fuels very well.

What is wrong with a rolling road?

Other people have asked why not use a rolling road? The dynamometer at MACE provides the ultimate test cell for an engine. The engine is fully accessible allowing a range of thermostats, vacuum gauges and exhaust gas monitors etc. to be fitted. In addition, it is easy to change the fuel that is being used. The throttle setting can be fixed and the revs controlled by the dynamometer, enabling part throttle as well as full throttle conditions to be investigated. This setup allowed a very large amount of data to be collected for many different scenarios.

The photo labeled "Engine and measuring equipment" shows the engine hiding behind a set of meters, with the air fuel ratio meter on the left-hand side and an array of eight yellow temperature meters connected to thermocouples on the fuel pump and carburettors. At the top right are the temperature readings for the water and exhaust gases. The rectangular blocks on top of the carburettors are to allow the height of the piston to be measured



Dynamometer

Engine and measuring equipment.

and you can just see one of the vacuum gauges connected to the inlet manifold to the left of the rear carburettor.

The photo above shows the water braked dynamometer with its large RPM gauge. This was managed from the control room where the gas analyser and various other readouts were located.

As the photos show, there is a lot more you can do with an engine installed in a proper test cell than you can with a rolling road.

What is the cause of our problems?

Modern fuel is different from classic petrol in two main ways. First, it most probably contains ethanol and second, it is made from of a much wider range of hydrocarbons.

Crude oil consists of many different hydrocarbon molecules. When heated, these molecules boil off at different temperatures, lighter ones evaporating first and the heavier ones at higher temperatures. Crude oil is cracked by heating and condensing out the components or distillates over the different temperature ranges. For example, light gases such as propane and butane are condensed at lower temperatures, petrol at higher temperatures, diesel at higher temperatures still and tars at the highest temperatures.

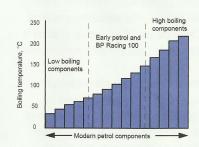
With a wide range of distillates modern petrol starts to evaporate

or boil at lower temperatures than classic petrol and the high boiling point components require higher temperatures to evaporate.

Before petrol can burn in the cylinder of a car, it must be a vapour. While the low temperature distillates make starting a cold engine easier, they are the sole cause of the "Hot restart problem". Some of these evaporate well below the operating temperatures of engines and under some circumstances will boil in the fuel system and carburettors, stopping the engine running properly.

In contrast, there may be insufficient heat generated when a low compression engine is running at a light throttle setting to evaporate the higher temperature components

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Components of modern fuel (Source BP)



Control Room

during the compression stroke. This can lead to increased Cyclic Variability and, potentially, overheating problems which are outlined later.

This subject will be covered in more detail in a future article on engine performance.

Ethanol

In the UK up to 5% ethanol alcohol is blended with petrol. In France this can be up to 10%, with even higher levels in other countries.

Adding ethanol to petrol is not new. Cleveland Discol was introduced in 1928 and sold until 1968, claiming to "contribute to a brilliant performance and better mileage because it keeps the engines cooler and cleaner", "the perfect cold-weather fuel". However, it is not known what percentage ethanol Cleveland Discol contained, making it difficult to compare with modern fuels.

Were Cleveland's claims true?

An E10 blend of fuel obtained in France was tested in Manchester and the report of how the XPAG ran on this fuel will appear in the engine performance article.



Corrosion in float chamber

There is a large amount of published information about ethanol blended fuels and their potential damaging effects on fuel hoses, seals, etc. However, probably the most worrying problem is their ability to corrode metals such as steel and aluminium.

Hoses are relatively cheap and easy to replace; carburettors, fuel pumps and the like, far more expensive and for older cars, parts may no longer be available.

Is corrosion of metal components really a problem?

Yes. You can see the corrosion at the bottom of one of the float chambers in my TC. Especially worrying as I have been using premium blend fuels that I thought were ethanol-free fuels for the past six years.

There are two different types of corrosion: one where the metal is attacked usually by an acid; the second requires two different, electrically connected, metals in a conducting medium, called an electrolyte. This arrangement forms a battery and as the current flows, the anode corrodes. This is called galvanic corrosion.

The FHBVC have tested and recommended additives that protect against acid corrosion; however, ethanol blended fuels also act as an electrolyte. The corrosion in my float chamber was around the steel bolt connected to the aluminium body, suggesting it is galvanic corrosion. There is little information on the protection offered by the additives against galvanic corrosion and is something I will investigate further this winter.

Do classic cars run as well on modern fuel?

How many times at Natters have you heard statements such as "modern fuel burns more slowly" or "the higher the RON number, the faster the fuel burns" or "modern fuel burns hotter"? While none of these statements are true, they

underlie a feeling that classic cars do not run as well on modern fuel.

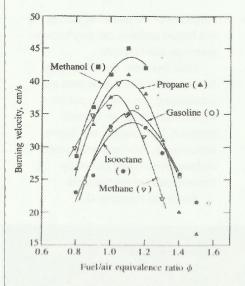
The way petrol burns in the cylinder is very complex. An overview of this process helps in understanding why classic cars can run hotter and why modern fuel apparently burns more slowly.

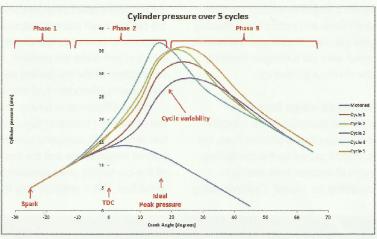
The graph, from a research paper, shows the burning velocity (flame front speed) of different hydrocarbons plotted against the fuel air equivalence ratio. Air equivalence ratio is defined as one when there are the exact number of oxygen molecules in the mixture to allow each of the hydrogen and carbon atoms in the fuel to oxidise or burn. Optimum power is produced at an air/fuel ratio of around 0.95 (or 1.05 on the graph as this shows fuel/air ratio).

While the flame front speed in different hydrocarbons is small, it is very sensitive to the differences in air/ fuel ratio. What is also interesting to note is the fastest flame front speed in Gasoline (Petrol) is a very sedentary 35cm/sec.

The bore of an XPAG engine is 6.67cm. At a speed of 35cm/s, it would take 0.2 seconds for the flame front to cross from the spark plug to the other side of the cylinder. If this were the only factor in burning the fuel, it would limit the engine to a maximum of 50rpm! The only reason spark ignition engines can run at high revs is because turbulence in the gases mixes the flame front in the cylinder allowing the burning to spread out more quickly. The turbulence of the air/ fuel mixture in the engine is a function of the design of the head and inlet manifold and not the type of fuel being used.

There are three phases during the combustion cycle. First, when the spark plug fires, a fireball of burning mixture about the size of a pin head is created. This fireball grows as the flame front moves outward at approximately 35cm/sec depending





Three phases of combustion

Research paper graph

on the pressure in the cylinder and air/ fuel mixture around the spark plug. This initial phase of the combustion is slow and a significant factor in the time taken to burn the fuel. Once the fireball has grown to approximately the size of a pea, the second phase begins when turbulence starts to take over and spreads these ignition points throughout the volume of the cylinder, rapidly igniting the remaining mixture and raising the pressure in the cylinder. Finally, any fuel which did not initially evaporate or was trapped around the valves or piston burns in the extremely high temperatures created during the second phase.

The time it takes from the spark that creates the initial fireball to all the mixture being burned is approximately constant and independent of engine revs. As revs increase it is necessary to advance when the spark plug fires to provide sufficient time for the fuel to burn before the optimum when the piston is approximately 17 degrees after top dead centre. On very early cars this was achieved manually, on later cars, and the majority of MGs, this is done by bob weights in the distributor which fly out as engine revs increase, causing the ignition timing to advance.

There is also a second effect. The growth of the initial fireball is dependent on the pressure of the mixture in the cylinder which in turn depends on throttle setting. At light throttle settings, cylinder pressure is low and the growth of the flame front slower. This requires the timing of the spark to be further advanced for light throttle settings. On later cars this is achieved using the vacuum advance pod on the distributor which is connected to the inlet manifold. A light throttle setting reduces the pressure in the inlet manifold causing the pod to advance the ignition timing. Earlier cars do not have a vacuum advance.

The tests at Manchester have shown how important correct ignition advance is in allowing the engine to run cooler.

Cyclic Variability

A weak or rich mixture around the spark plug slows the growth of the initial fireball. This can have a significant effect on the timing of combustion cycle. Even if the carburettor is set to deliver the perfect fuel/air mixture, there is no guarantee that, after the compression stroke, the mixture around the spark plug is correct. It is quite possible it will be either too weak or too rich, depending

on how well the fuel is atomised in the carburettor, mixed with the incoming air and vaporised during the compression stroke. A slow growth of the initial fireball leads to retarded combustion of that cycle. The graph above are the results of cylinder pressure measurements in a running engine and show a difference of some 10° difference between the timing of the peak pressure.

Cycle by cycle variations in the mixture of the small volume of gases around the plug when it fires and subsequent changes to the speed at which the initial fireball grows, lead to a phenomenon called Cyclic Variability. The timing of each combustion cycle varies every time that cylinder fires. Even with a perfectly tuned engine with the correct centrifugal and vacuum advance, cyclic variability causes a percentage of the combustion cycles to burn too slowly as though the engine is running retarded, increasing the temperature of the hot gases leaving the exhaust.

The other effects of cyclic variability are that it causes an engine to run roughly and slightly reduces power output, normally something most people will not notice. However, large cyclic variability will noticeably increase the temperature of the exhaust gases and under-bonnet temperatures. It is

worth trying different fuel suppliers and blends and noting how smoothly your car runs. Should you find one on which your car runs more smoothly, use that as a preference.

Where do we go from here?

With the exception of the corrosion caused by ethanol blended fuels, the problems caused by the low boiling point of modern fuels and cyclic variability are related. Cyclic variability

leads to more heat being generated in the cylinder head and exhaust which raises under-bonnet temperatures, which in turn make the problem vaporisation of the low boiling point components of modern fuel worse.

Slight differences in tuning, and use of different brands of fuel, etc can reduce cyclic variability, underbonnet temperatures and the impact of the low boiling point components in the fuel. This is probably why the

symptoms of the problems, such as the Hot Restart problem, can vary between seemingly identical cars.

Initial tests with my TC using the findings of the Manchester tests are encouraging. The aim is to publish three future articles on: Ethanol corrosion, Fuel volatility, and Engine performance, with recommendations on how the problems caused by modern fuels can be mitigated.

Watch this magazine.

Paul Ireland - Personal Profile

Paul was born in the early 1950s and bought his first car, a 1949 MG TC, for £60 in 1967. With the help of his father, the body was removed and suspension, wheels, steering and brakes refurbished and he rewired it with a home-made loom. New wood in the sills and a brush coat of paint made the car roadworthy. With no money to rebuild the engine, Paul rattled and burned oil around Manchester University in his TC. In the early 1970s student cars were a rarity and owning a classic MG, no matter what condition, was a real status symbol.

With a PhD in experimental Nuclear Physics, Paul is the black sheep of the family; both grandfathers, his father and two sons are proper engineers.





The TC when bought in 1967

However, this experience has allowed him to take a more academic approach to the problems he has running his TC on modern fuel and enabled him to gain a better understanding of the problems for the benefit of all.

After his chassis-up restoration in 2003, Paul has shown his TC and used it for tours and longer trips, and for testing different fuel mixes.

